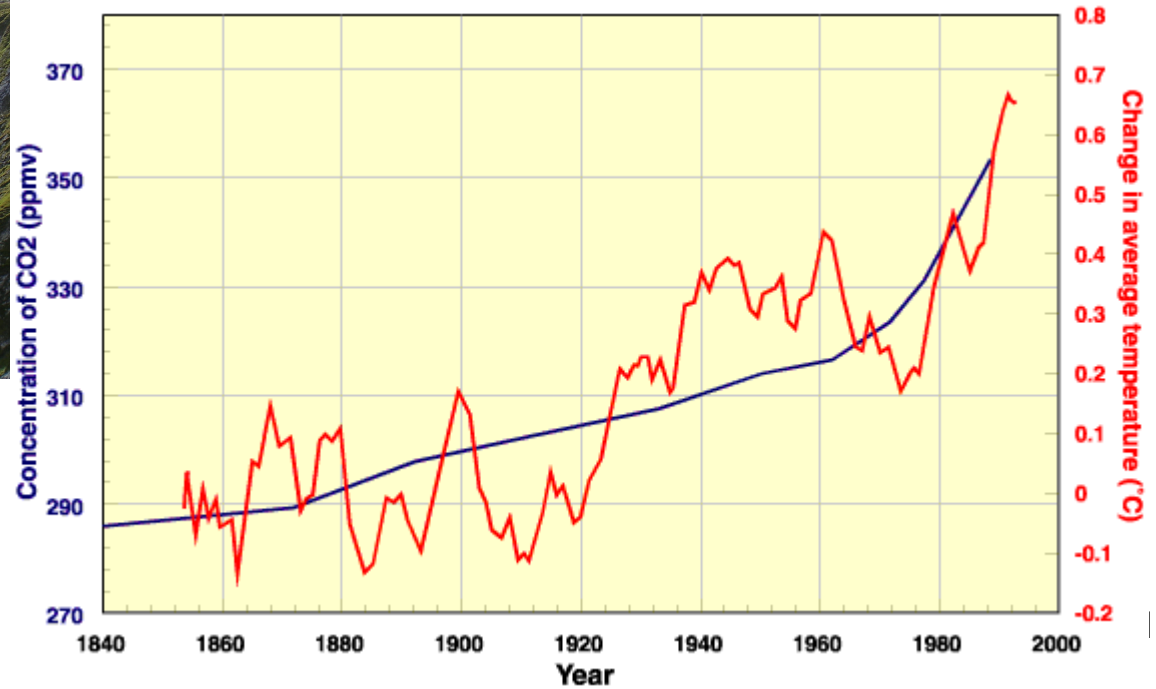
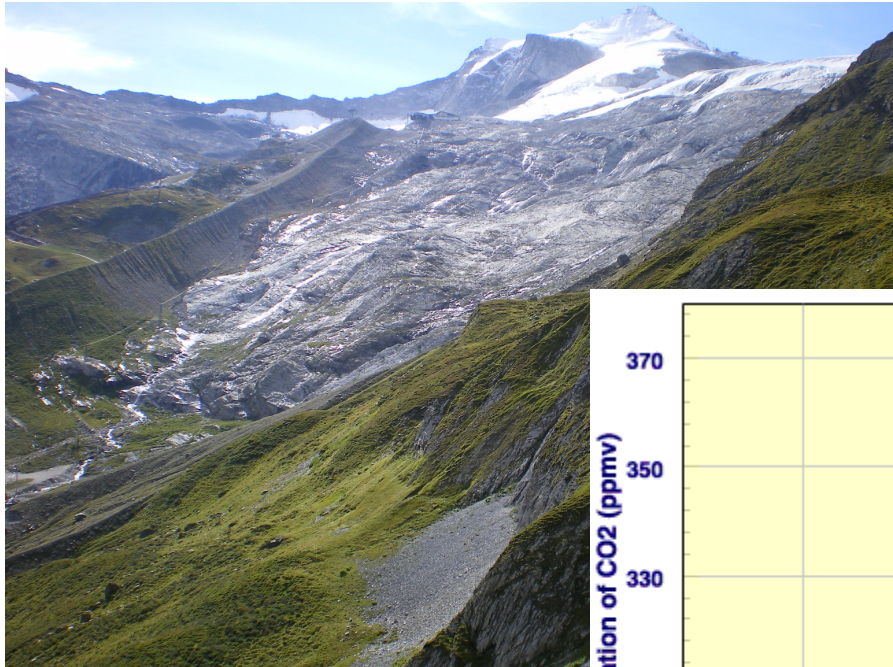
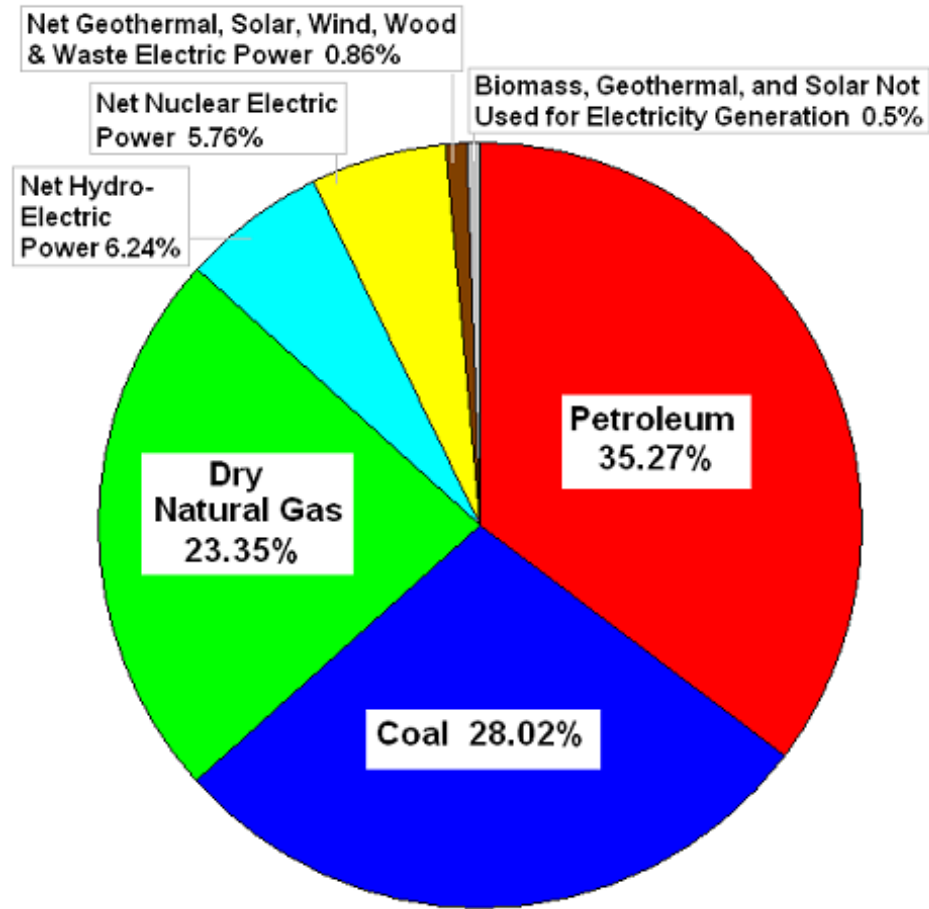


# Global warming and greenhouse effect

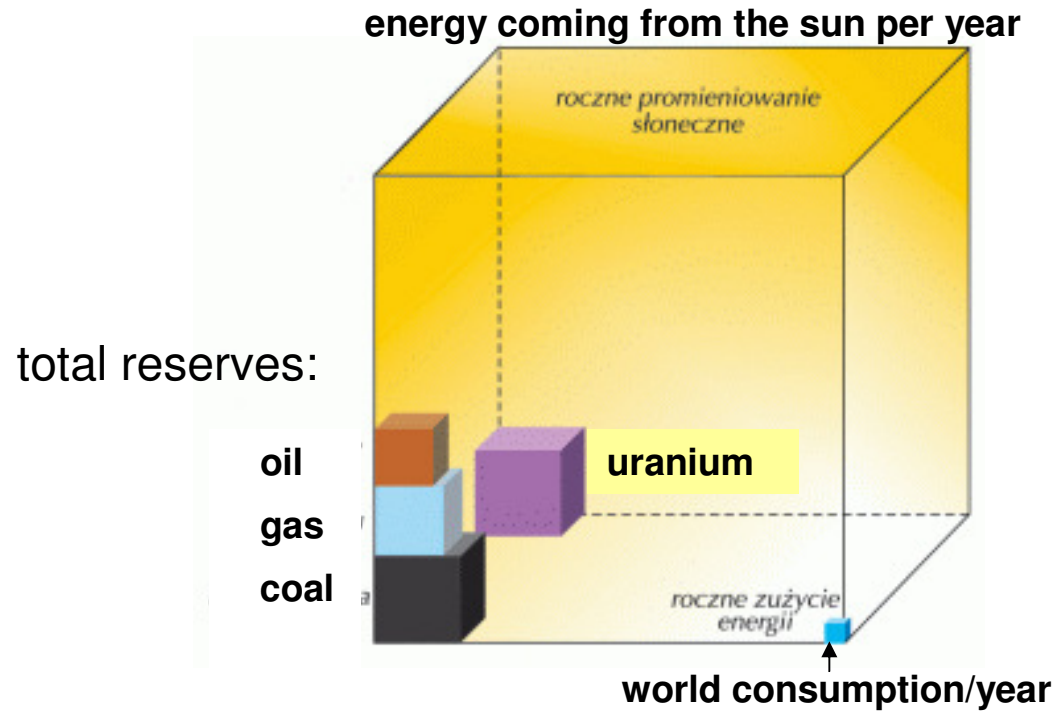
## Alps 2007





World Consumption 2006

# Energetical resources: fossil and nuclear fuels compared to the energy from the Sun



## Renewable energy

- ☐ hydro 5 TW= $5 \times 10^{12}$  W
- ☐ geotherm. – up to 20 TW
- ☐ wind – 50 TW (27 % area of Earth)
- ☐ biomass 20 TW (31% area of Earth)
- ☐ sun 600 TW

*Natan Lewis, Caltech*

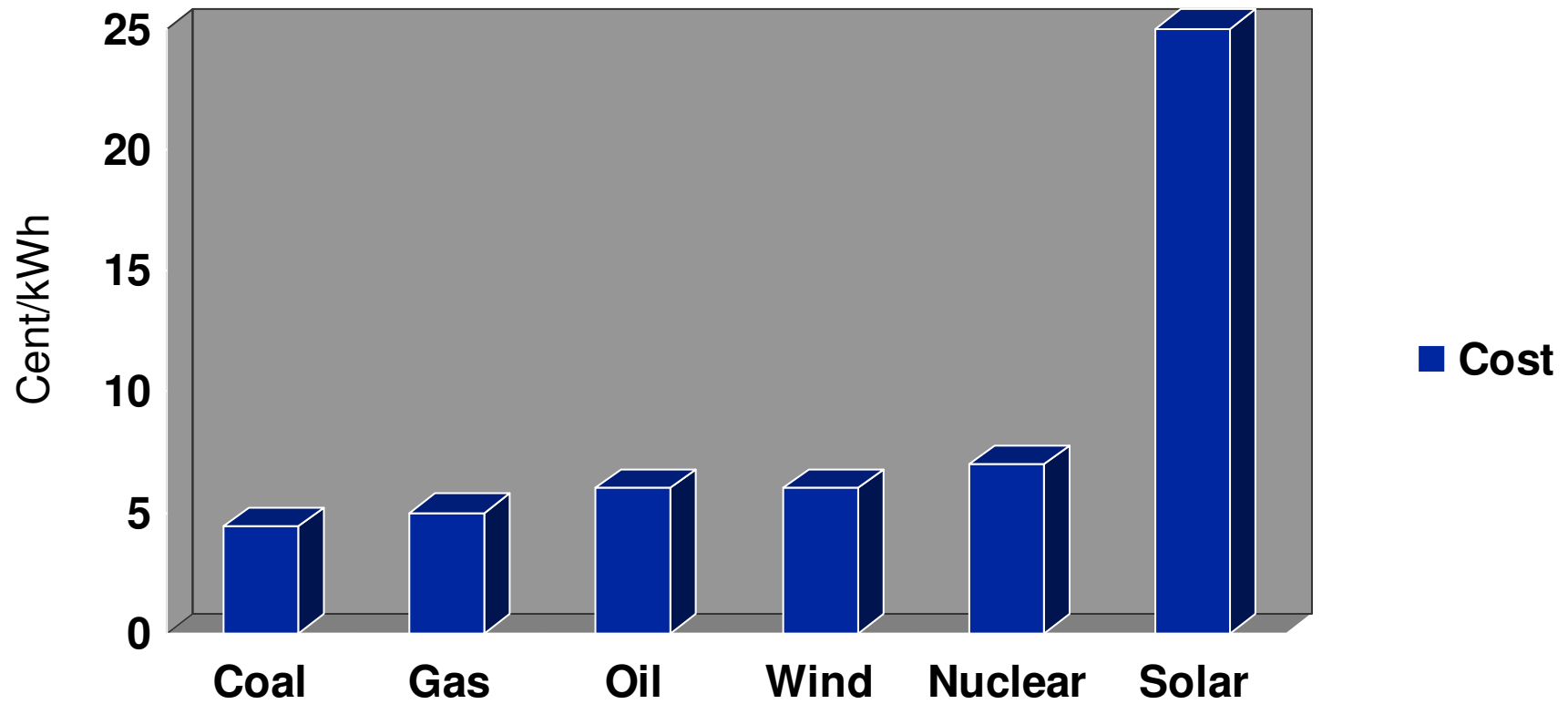
Present and expected energy demand

2010: 15 TW

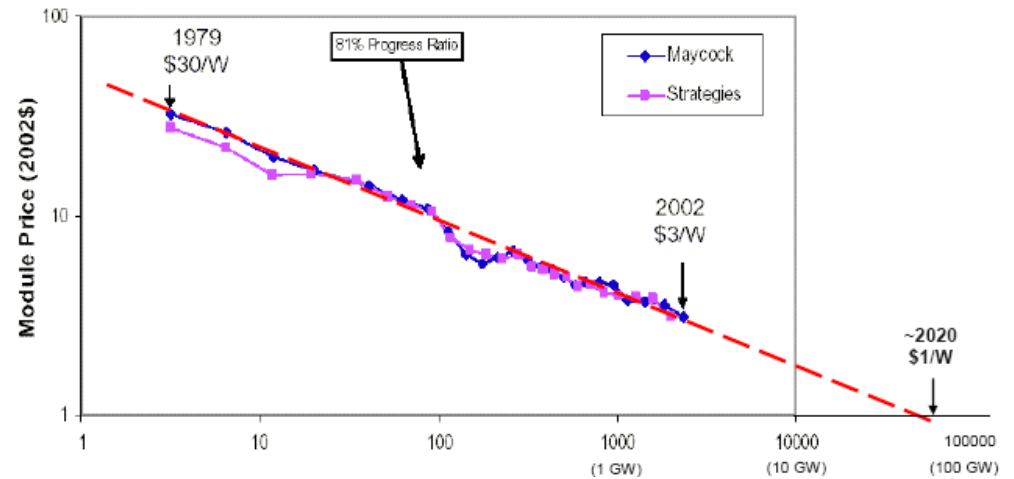
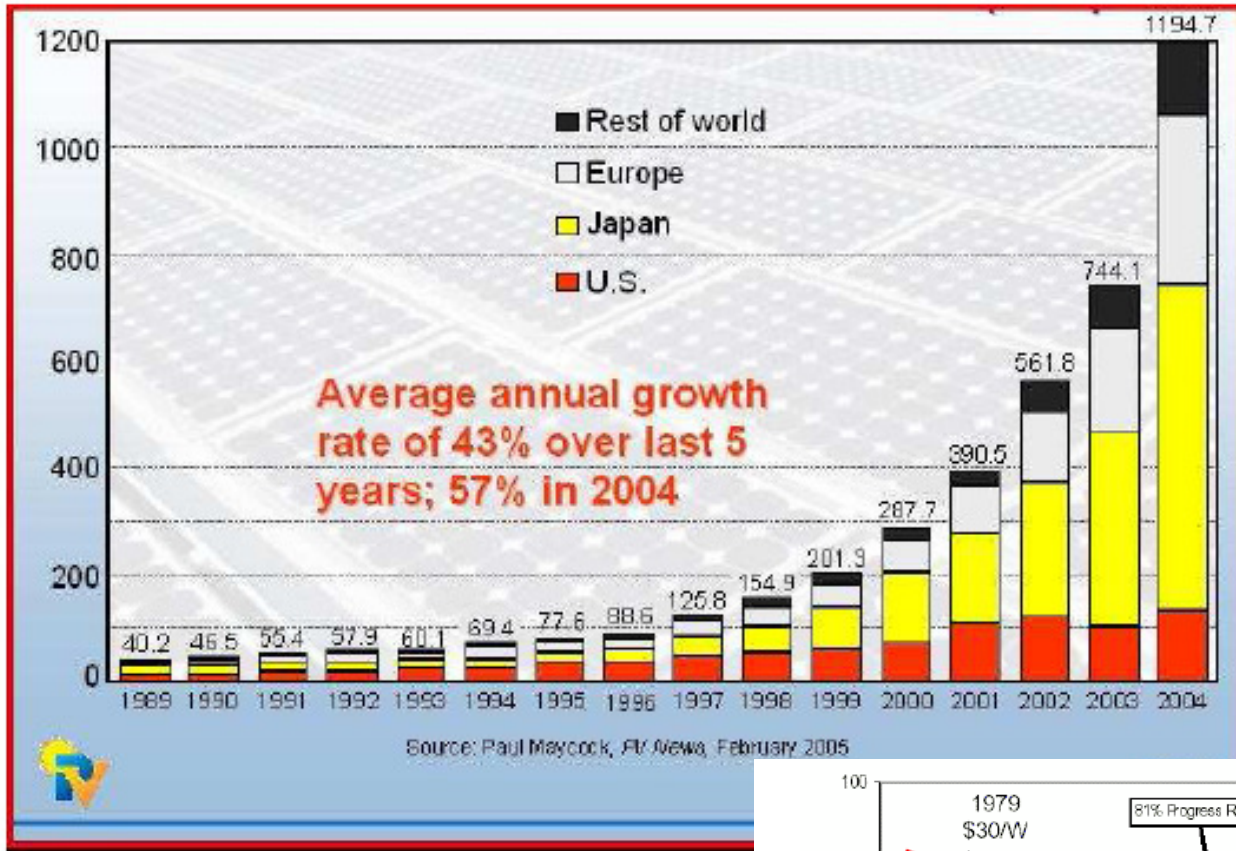
2050: 28 TW

*generation of 20 TW at 10 % eff.  
requires 8.8 % area of USA*

# Electricity - production costs (USA, 2002)



# Growth of photovoltaic industry



# Photovoltaics – energy from the sun



## Topics

1. The characteristics of sunlight
2. Electrons and holes in semiconductors: band model, doping, current transport, absorption of light, defects and recombination
3. Junctions: pn, heterojunction
4. Principles of solar cell: photovoltaic effect, conversion efficiency, basic design, efficiency limits and photovoltaic losses
5. Simulation of solar cells performance
6. Monocrystalline solar cells: silicon, GaAs
7. Thin film solar cells: amorphous silicon, heterojunction cells (CIGS, CdTe)
8. Other concepts: Graetzel cell, organic etc
9. 3<sup>rd</sup> generation photovoltaics, new ideas
10. Modules: design, problems & solutions
11. Light management: concentration, light confinement
12. Photovoltaic systems: stand-alone, grid-connected, concentrator



## Literature

J. Nelson „The physics of solar cells”

S.R.Wenham „Applied photovoltaics”

R.H. Bube „Photovoltaic materials”

A. Rockett „The materials science of semiconductors”

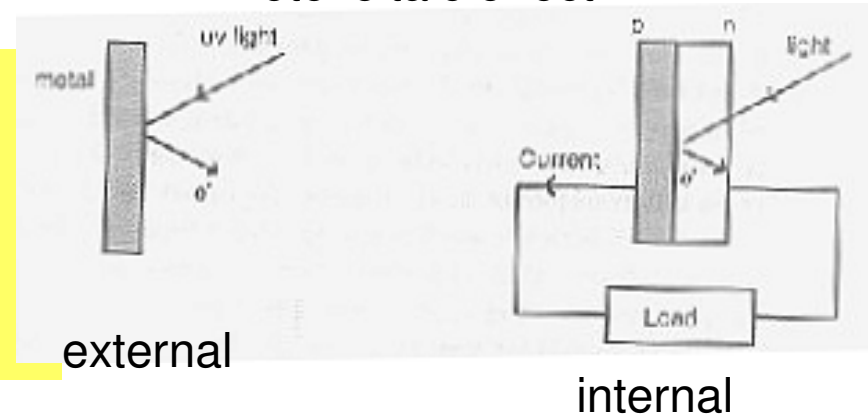
P. Würfel „Physics of solar cells”

PV CDROM

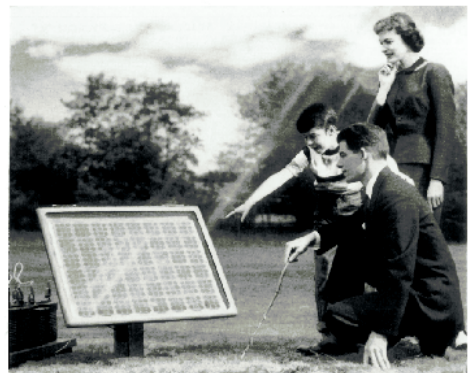
# History

## Photovoltaic effect

- Becquerel 1834 electrolyte
- Adams & Day 1877 Se
- Chapin, Fuller, Pearson 1954 Si 6%
- CdS pn junction 6% 1954
- Vanguard 1, 1958



## The First Practical Solar Cell-1954



**Something New Under the Sun.** At the Bell Telephone Laboratories, one of the first solar cells is specially trained to run on impurities of germanium. It converts the sun's rays directly into a sufficient amount of electricity. Simple and mobile fuel. (The energy batteries inside the solar battery store up its electricity for night use.)

### Bell System Solar Battery Converts Sun's Rays into Electricity!

*Bell Telephone Laboratories invention has great possibilities for telephone service and for all mankind*

Ever since Archimedes, men have been searching for the secret of the sun.

For it is known that the sun's rays help the flowers and the grains and the fruits to grow also send an almost limitless power. It is nearly as much every three days as in all known reserves of coal, oil and uranium.

If this energy could be put to use—there would be enough to turn every wheel and light

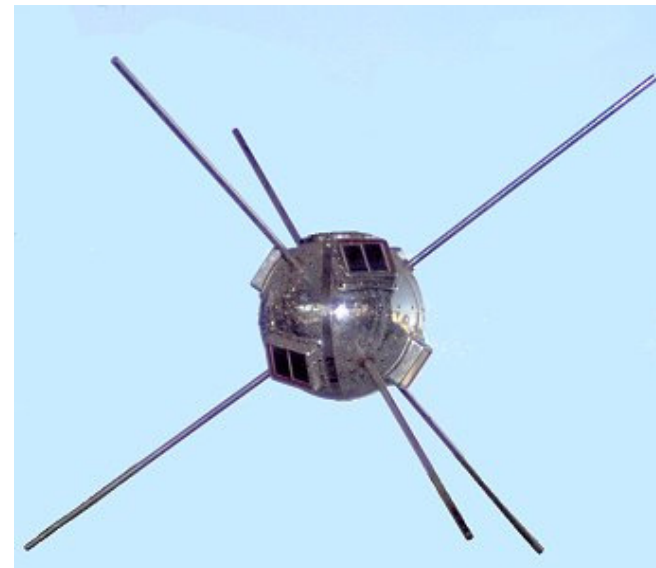
long research and first announced in 1954. Since then its efficiency has been doubled and its use is now a reality.

There's still much to be done before the battery's possibilities in telephony and for other uses are fully developed. But a good and pioneering start has been made.

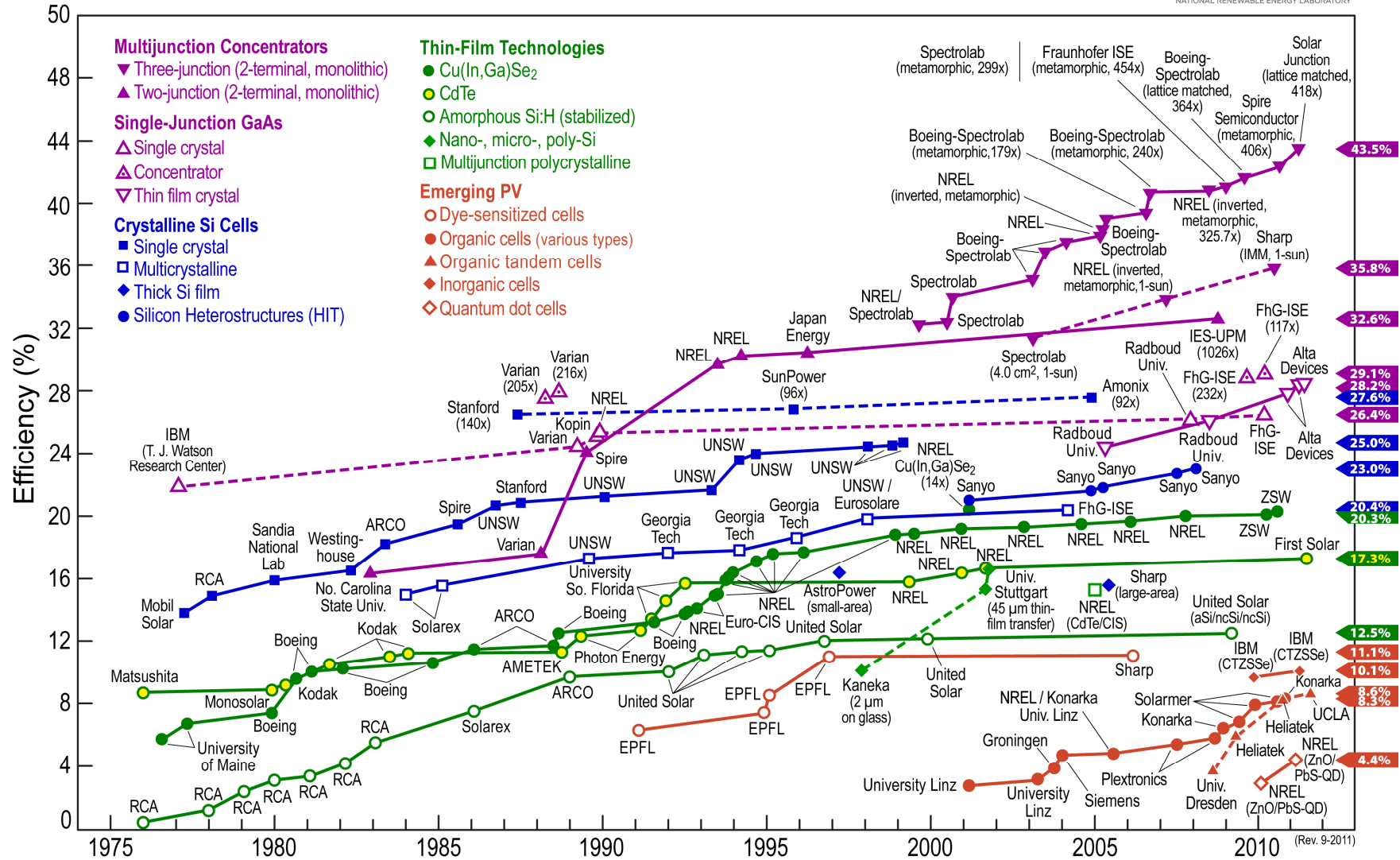
The progress so far is like the opening of a door through which we can glimpse exciting

ENIC 6/24/06  
H. Atwater Caltech

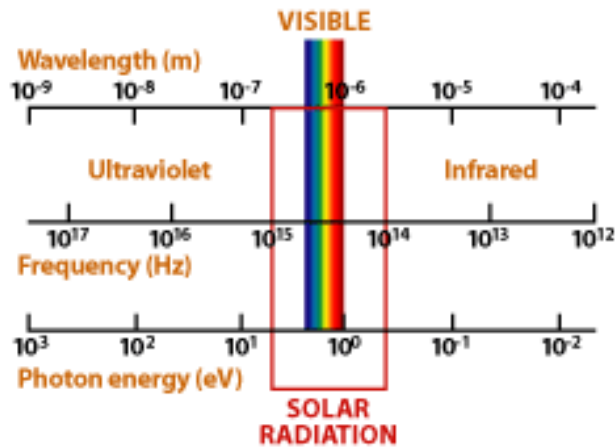
## Vanguard, 1st terrestrial satellite, 1958



# Best Research-Cell Efficiencies



# Electromagnetic radiation



AM - air mass

AM0: just outside the atmosphere

1.3661 kW/m<sup>2</sup> (solar constant)

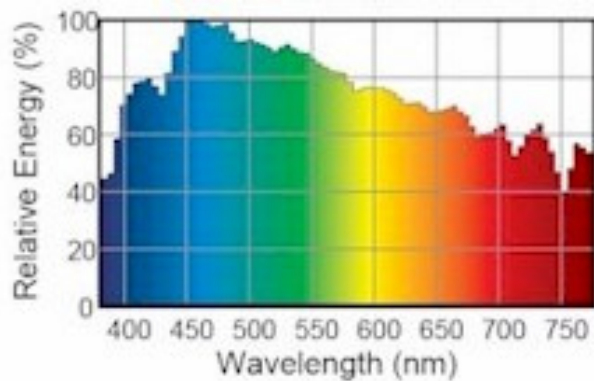
Per year: 1.188 kWh/cm<sup>2</sup>

AM1=after 1 atmosphere thickness

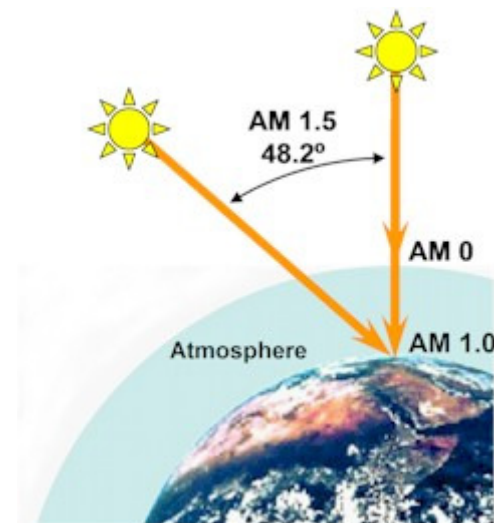
$AMX = AM1 / \cos\phi$

$A.M1.5 \cong 1 \text{ kW/m}^2 \quad \phi \cong 48.2^\circ$

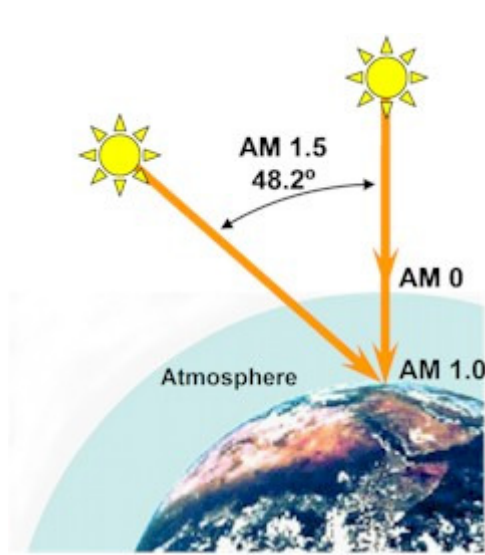
(it corresponds to  $\sim 42^\circ$  latitude on equinox)



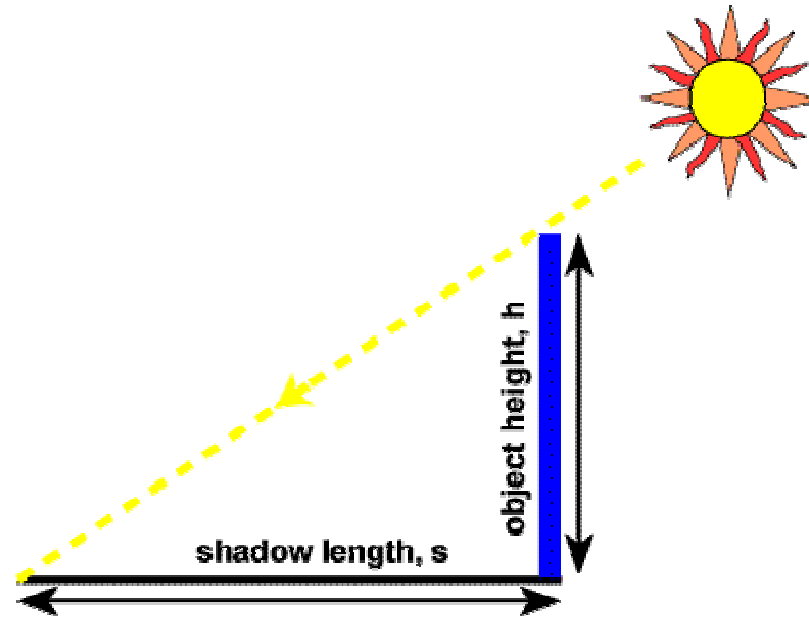
Solar spectrum in visible light



# How to measure Air Mass coefficient

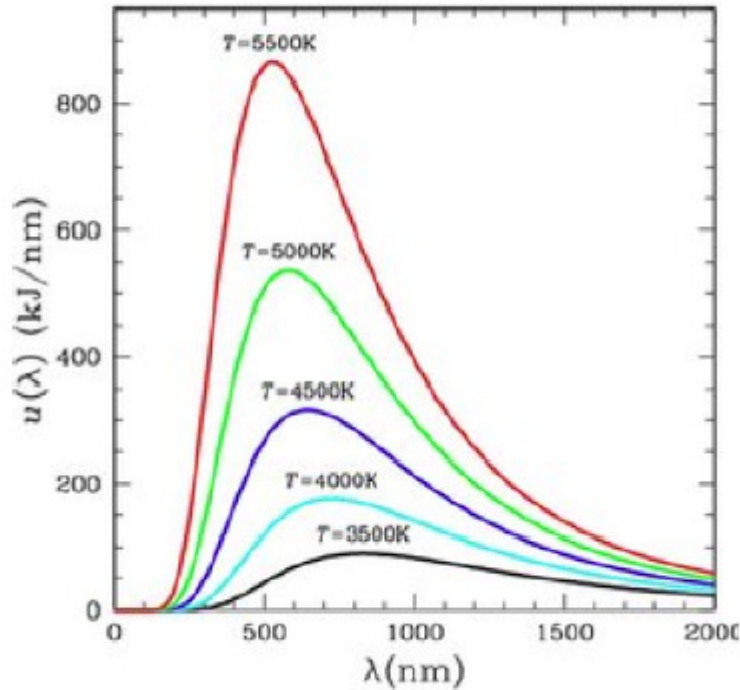


$$AM_X = AM_1 / \cos \phi$$



$$AM = \sqrt{1 + \left(\frac{s}{h}\right)^2}$$

# Black-body radiation



Planck's distribution

$$E(\lambda, T) = \frac{2\pi hc^2}{\lambda^5 [e^{\frac{hc}{\lambda kT}} - 1]}$$

photons:  
 $E = h\nu = hc/\lambda$   
 $\lambda = c/\nu$

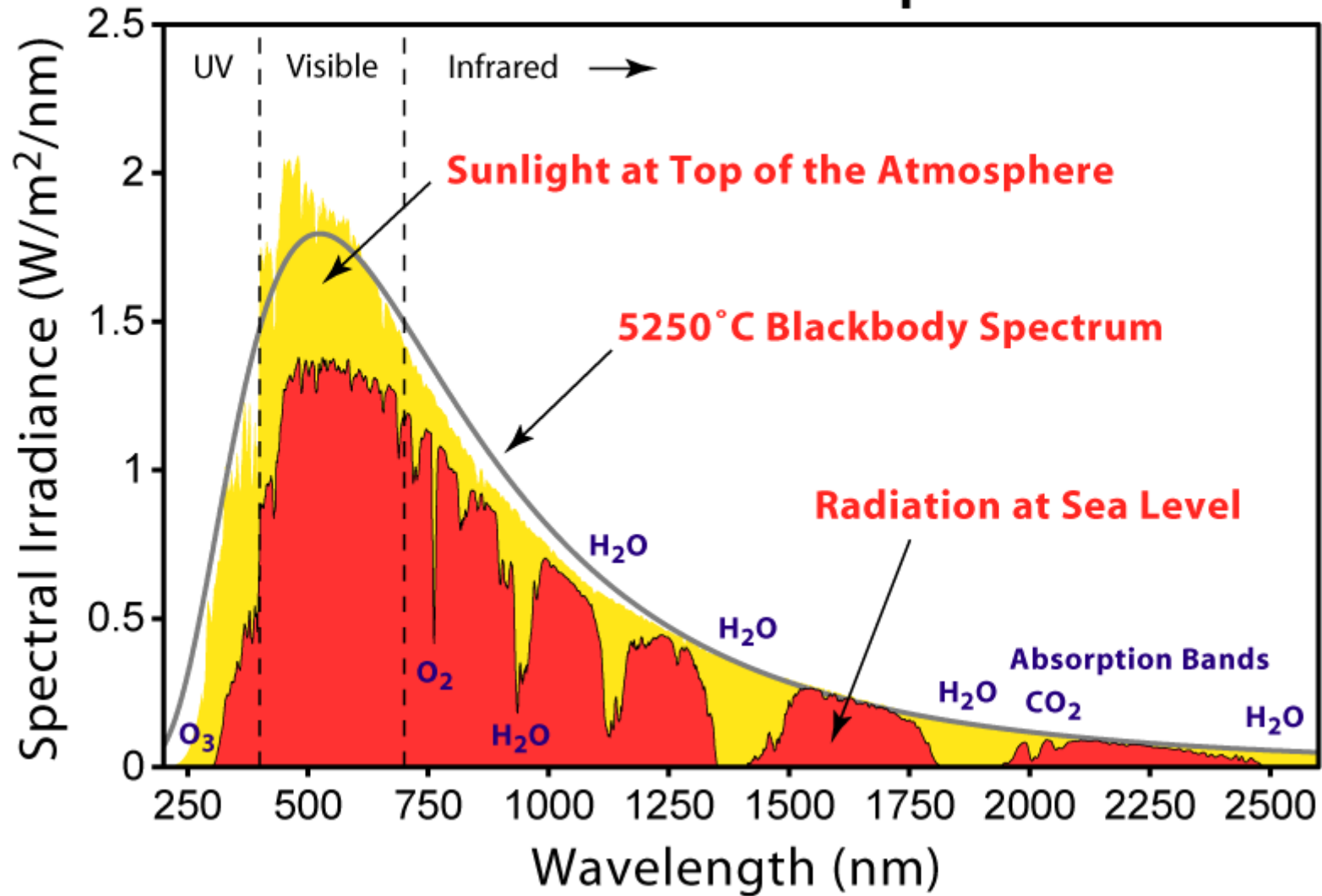
Stephan-Boltzmann law

$$\int_0^{\infty} E(\lambda, T) d\lambda = \sigma T^4 \quad [\text{W/m}^2]$$

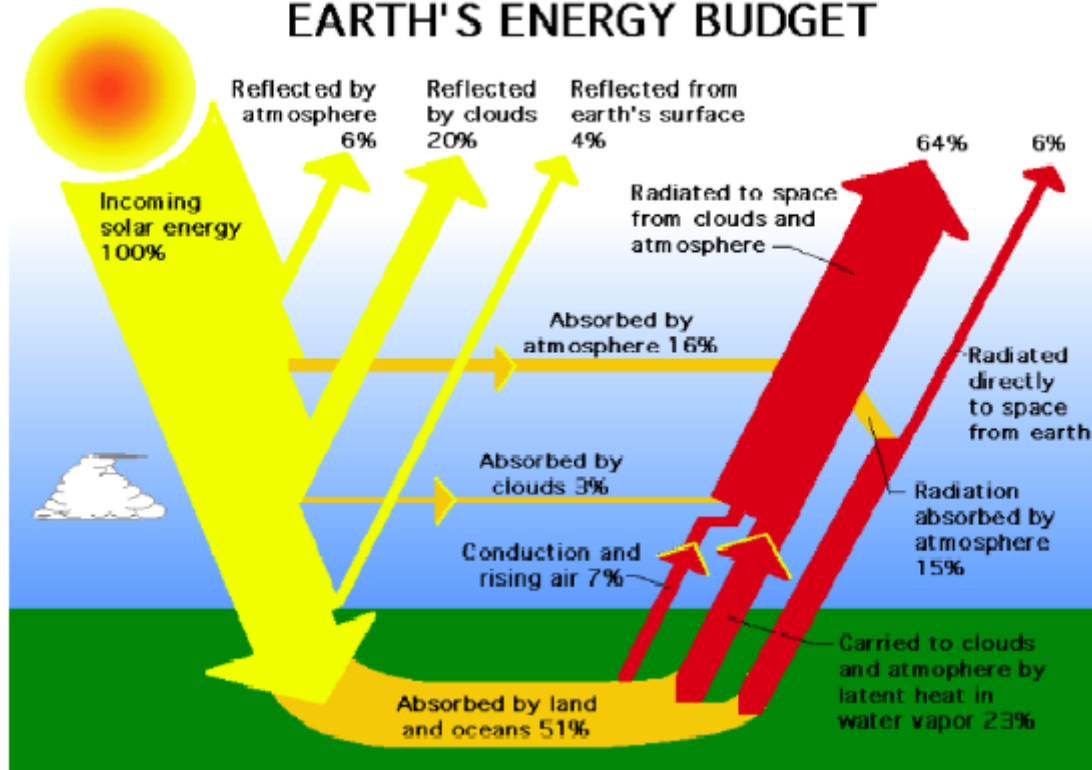
$h = 6.6 \times 10^{-34}$  Js Planck's constant  
 $k = 1.38 \times 10^{-23}$  J/K Boltzmann's constant

**Wien law:**  $\lambda_{\text{max}} = \frac{b}{T}$   
 $b = 2.9 \cdot 10^{-3}$  mK

# Solar Radiation Spectrum

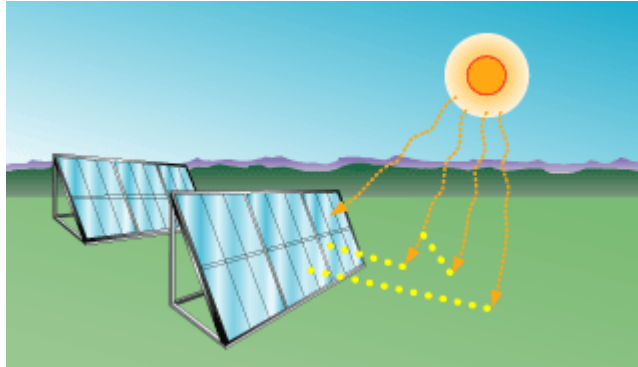


# EARTH'S ENERGY BUDGET

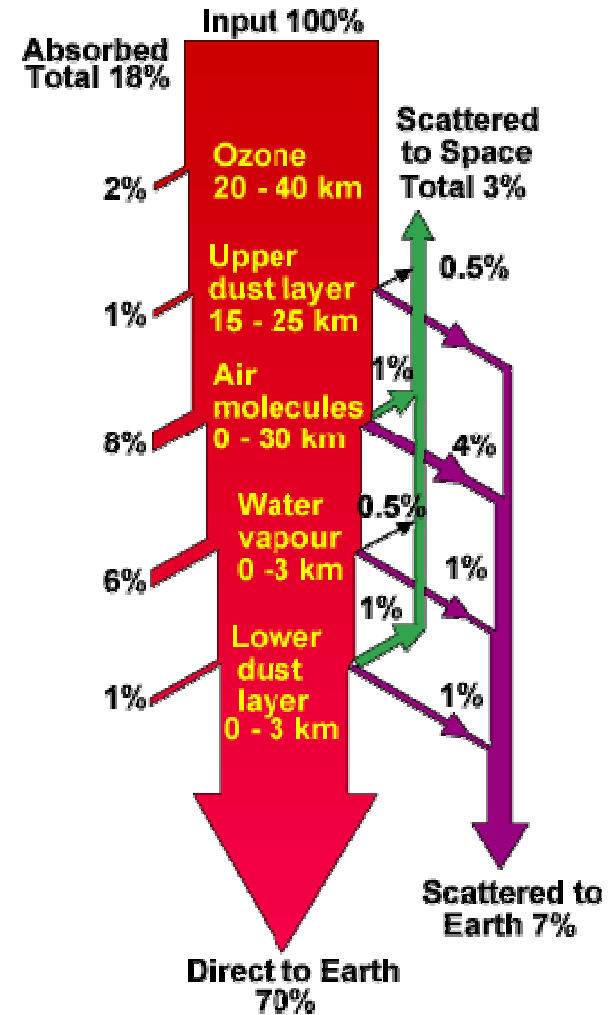
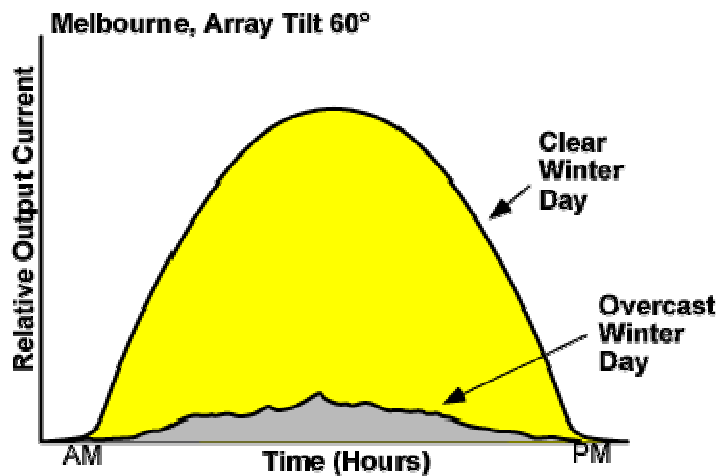




# Direct and Diffuse radiation



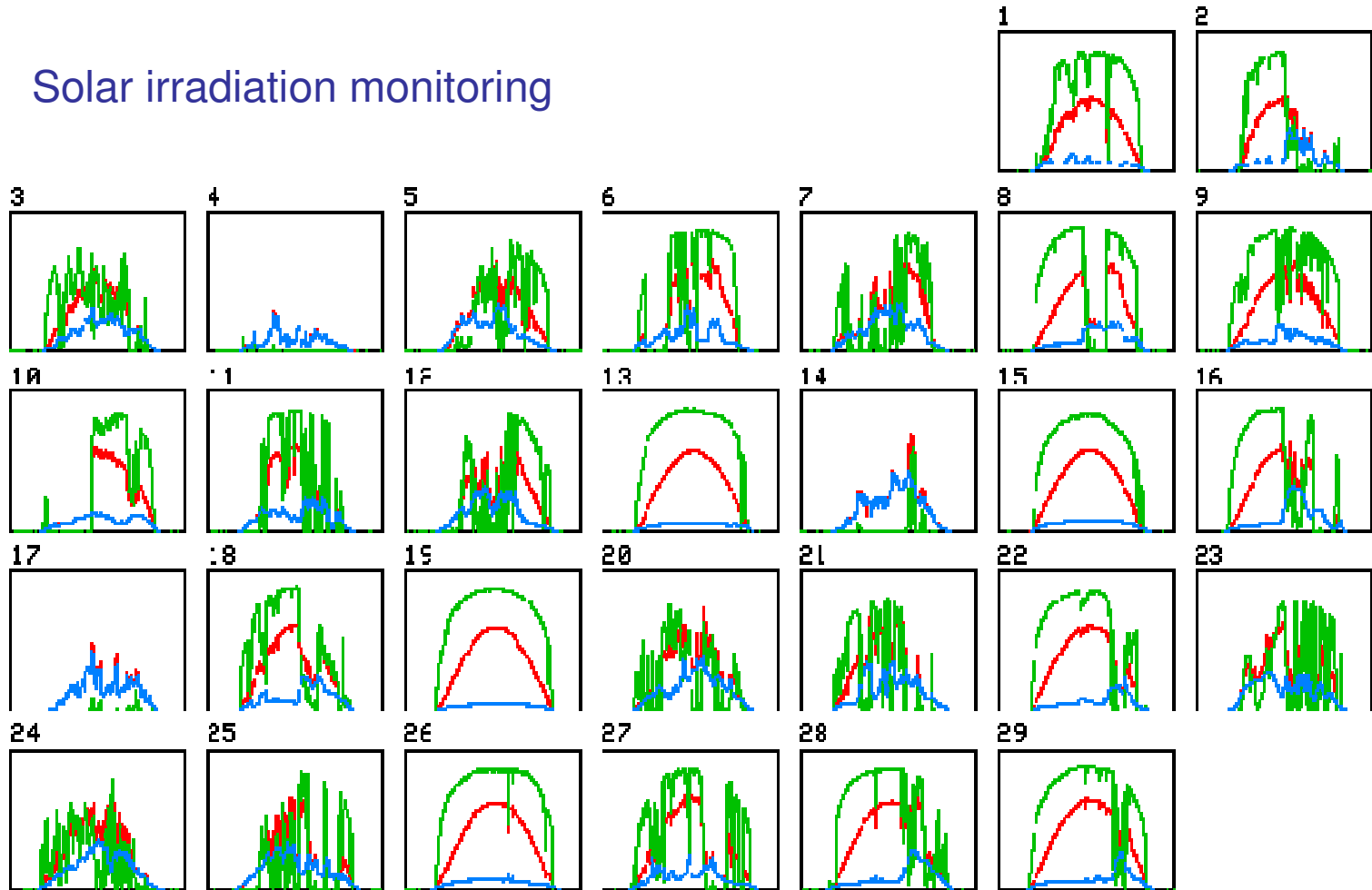
10% diffuse component at clear skies, AM1



Typical clear sky absorption and scattering of incident sunlight <sup>17</sup>

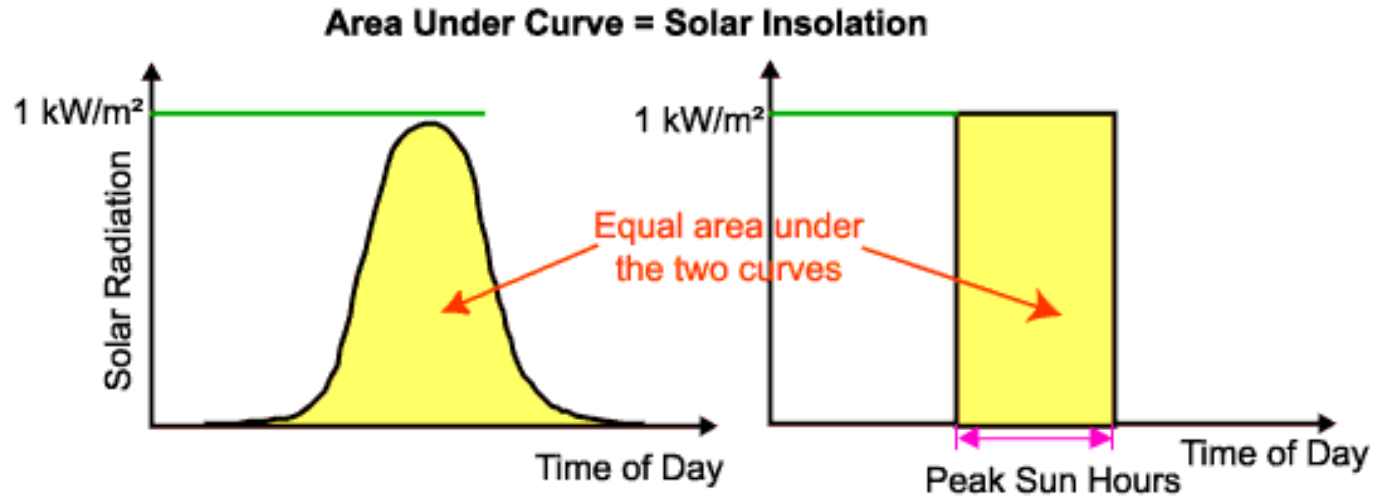
**Solar Radiation Research Laboratory (BMS)**  
**February 2008 Solar Calendar (NREL)**

Solar irradiation monitoring



Red = Global, Green = Direct, Blue = Diffuse

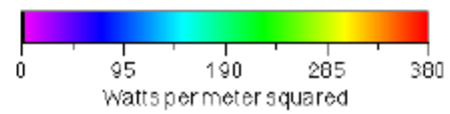
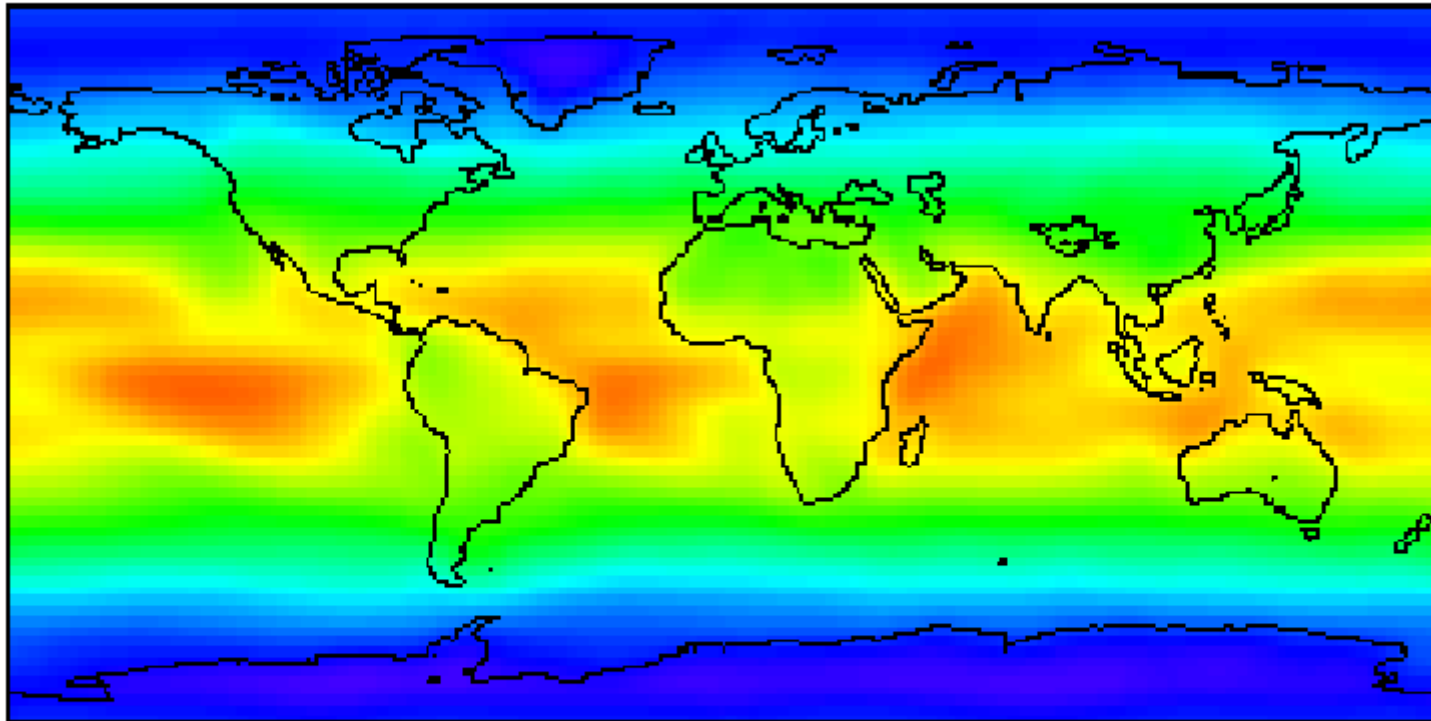
# Peak Sun Hours



*Peak sun hours = solar insolation if the sun were shining at its maximum value for a certain number of hours*

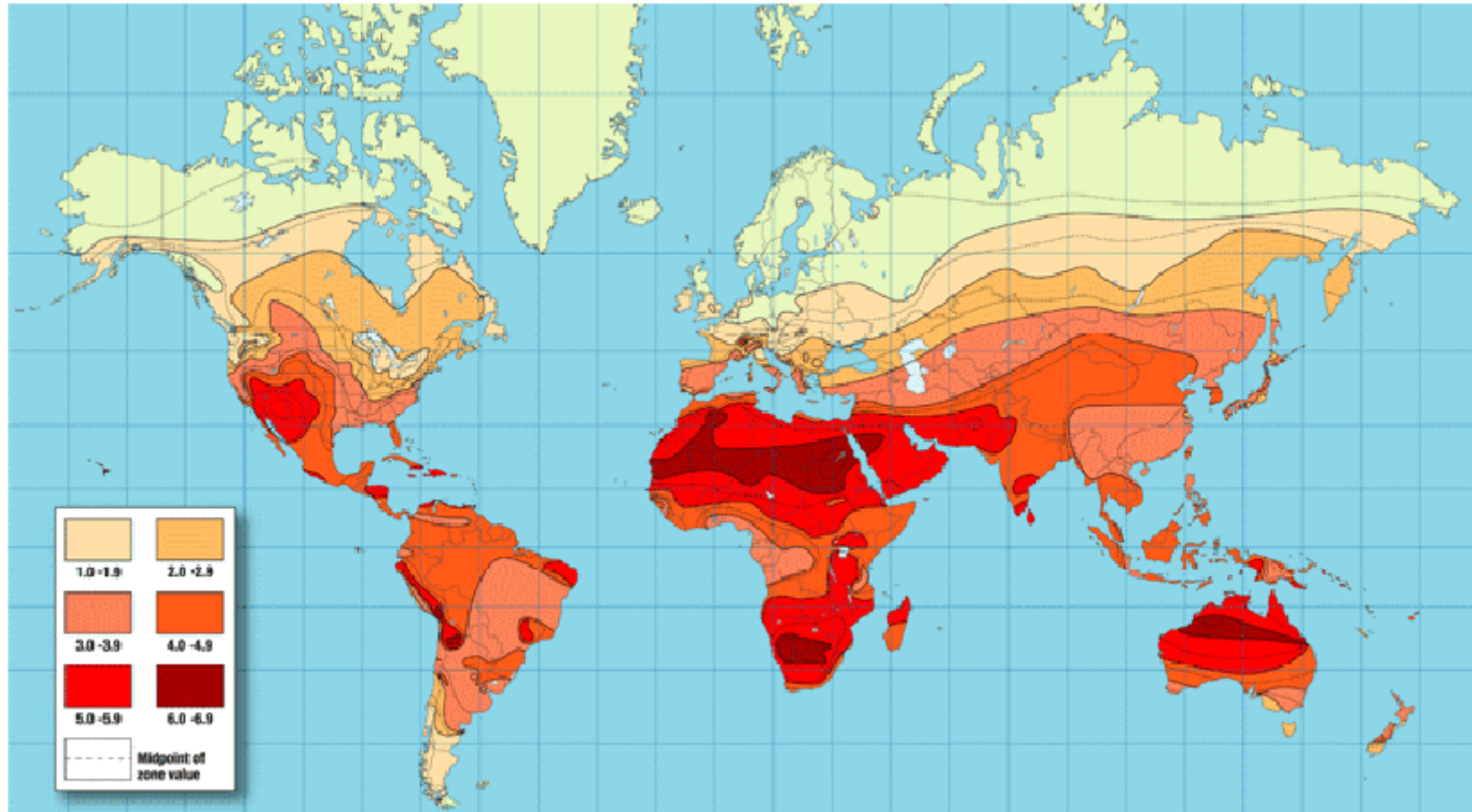
8 kWh/m<sup>2</sup> per day = 8 hours of sun at 1 kW/m<sup>2</sup> per day

## Average insolation intensity

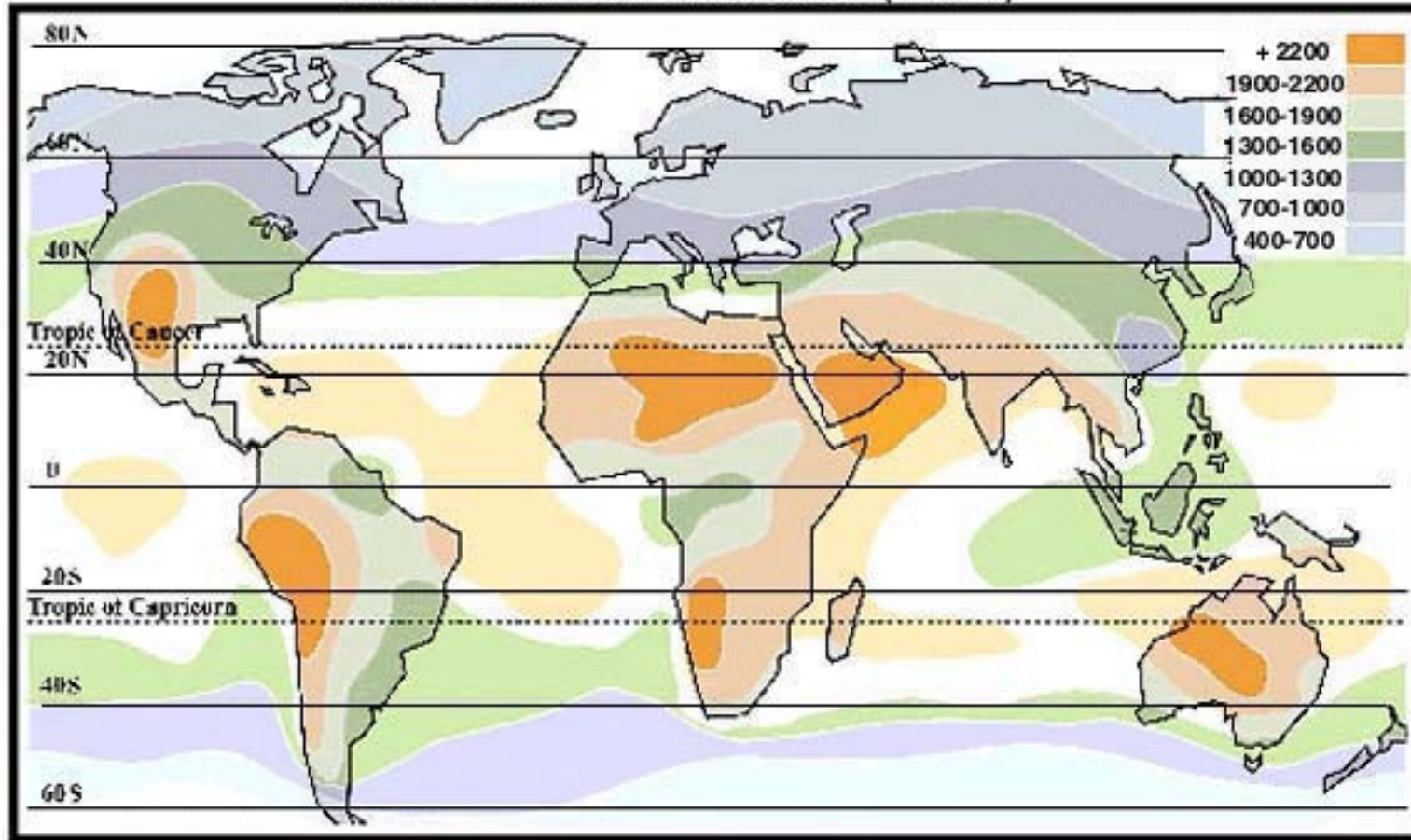


# World insolation chart – energy (kWh/m<sup>2</sup>) during winter day

*peak sun hour data= total daily insolation kWh/m<sup>2</sup>*



WORLD SOLAR ANNUAL RADIATION (kWh/m<sup>2</sup>)



# Annual Sum of Global Irradiance 2004

